School of Chemistry and Molecular Bioscience

CHEM102: Chemistry 1B: Structure and Reactivity of Molecules for Life

Subject Outline
Spring 2019
On-Campus
Wollongong

Subject Information
Credit Points: 6
Pre-requisite(s): CHEM101: Chemistry 1A, except with permission from Head of School
Co-requisite(s): Nil
Restrictions: Nil
Contact Hours: 3 x 1 hr Lecture; 1 x 1 hr Tutorial; 1 x 3 hr Practical

Subject Contacts
Subject Coordinator/Lecturer

<table>
<thead>
<tr>
<th>Name:</th>
<th>A/Prof Glennys O'Brien</th>
</tr>
</thead>
<tbody>
<tr>
<td>Location:</td>
<td>Building 18, Room 112</td>
</tr>
<tr>
<td>Telephone:</td>
<td>61 2 4221 3072</td>
</tr>
<tr>
<td>Email:</td>
<td><a href="mailto:glennys_obrien@uow.edu.au">glennys_obrien@uow.edu.au</a></td>
</tr>
<tr>
<td>Consultation mode and times:</td>
<td>Email for appointment</td>
</tr>
</tbody>
</table>

Student Support and Advice
For general enquiries please contact StudentHub 41:

Location: Bldg 41 Level 1
Telephone: 61 2 4221 3492
Email: smah-students@uow.edu.au
# Table of Contents

Section A: General Information ............................................................................................................... 3  
Subject Learning Outcomes .................................................................................................................. 3  
Subject Description ............................................................................................................................. 3  
Readings, References and Materials .................................................................................................. 3  
  Textbooks ........................................................................................................................................ 3  
  Prescribed Readings (includes eReadings) .................................................................................... 3  
  Materials .......................................................................................................................................... 3  
  Recommended Readings ................................................................................................................ 4  
Recent Changes to this Subject .......................................................................................................... 4  
List of Topics Covered ........................................................................................................................ 4  

Section B: Assessment ........................................................................................................................... 5  
Assessment Summary ........................................................................................................................ 5  
Details of Assessment Tasks .............................................................................................................. 5  
Minimum Requirements for a Pass in this Subject ............................................................................. 6  
  Minimum Student Attendance and Participation ............................................................................. 6  
Scaling................................................................................................................................................. 6  
Late Submission................................................................................................................................ 7  
  Late Submission Penalty – at 10% ................................................................................................. 7  
Supplementary Assessments ............................................................................................................ 7  
System of Referencing Used for Written Work ................................................................................... 7  
Submission of Assessments ............................................................................................................... 7  
Assessment Return ............................................................................................................................. 8  

Section C: General Advice ...................................................................................................................... 9  
Student Consultation and Communication .......................................................................................... 9  
eLearning Space ................................................................................................................................... 9  
Use of Internet Sources ...................................................................................................................... 9  
Lecture, Tutorial, Laboratory Times .................................................................................................. 10  
Extraordinary Changes for the Subject after Release of the Subject Outline ................................... 10  
Learning Analytics ............................................................................................................................ 10  
The Assessment Quality Cycle ......................................................................................................... 10  
Academic Integrity Policy .................................................................................................................. 10  
Student Academic Complaints Policy (Coursework or Higher Degree Research) ........................... 11  
Student Support Services and Facilities ........................................................................................... 11  
Student Etiquette ............................................................................................................................... 11  
UOW Grade Descriptors ................................................................................................................... 12  
University Policies ............................................................................................................................ 13  
Version Control Table ....................................................................................................................... 14
Section A: General Information

Subject Learning Outcomes

On completion of this subject, students should be able to:

1. Apply equilibrium concepts generally to aqueous solutions & their reactions in example applications;

2. Describe basic transition metals complex chemistry in bonding terms & describe example applications;

3. Outline inorganic chemistry of S, P & N, appropriately related to acid base & redox reactions & applied in biological or environmental contexts;

4. Describe electrochemical processes in terms of free energy, equilibrium & redox reactions;

5. Approach organic chemistry from viewpoints of structure, bonding, isomerism, & functional group reactivity using a variety of molecular representations;

6. Take a mechanistic approach to describe & analyse organic reactions types applying electron movement, bond breaking & bond making & stability concepts & using appropriate representations. This approach will be applied to understanding examples of biological reactions, including synthetic & biopolymers.

Subject Description

The subject follows on from CHEM101, using concepts and principles concerning the structure of matter and the nature of chemical change, applied first in aqueous reactions of acid base, and redox systems. A suite of inorganic compounds and reactions types including periodicity is then introduced covering transition metal complexes and inorganic non-metals, of S, P, N relevant to biological and environmental systems. Organic chemistry is specifically introduced with a detailed look at structure and isomerism leading to a treatment of organic functional groups and reactions based on mechanistic descriptions of the reaction systems. The mechanistic approach considers reactivity and stability of chemical species and uses thermodynamic, kinetic and equilibrium considerations to describe and analyse reactions. The application of structure concepts and reaction types is used to understand the properties of natural and synthetic polymers. For the detailed subject content, see subject content section within the CHEM102 subject handbook, and this content listing is also available on the subject Moodle site.

Readings, References and Materials

Textbooks

The following text(s) will need to be purchased by students enrolled in this class.


Prescribed Readings (includes eReadings)

The following readings are prescribed for this subject, but students are not expected to purchase these. They are available to students through the library on the subjects eLearning site.


Zeegers "Essential Skills for Science and Technology" Oxford University Press

Alternative First Year Chemistry texts can be found in the Library at 540.

Materials

1. The Subject Handbook containing details of the subject, laboratory instructions, templates for data collection and submission of lab reports and pre-labs as required. Pre-workshop/pre-lab study materials, are also in the handbook. The handbook is available for purchase from the UniShop. A copy of the handbook is available as pdf in Moodle.

2. Safety glasses, lab coats and closed footwear are mandatory in the laboratory. Lab coats and safety glasses may be purchased from the Unishop or any suitable supplier.
Recommended Readings
The following references complement the prescribed readings and textbooks:

Nil

Recommended readings are not intended as an exhaustive list, students should use the Library catalogue and databases to locate additional resources.

Recent Changes to this Subject
Nil

List of Topics Covered
The following are examples of the topics to be covered in this course. This is not an exhaustive list and will be subject to change.

1. Chemical equilibrium: Acid Base and solution equilibria including buffers.
2. Electrochemistry: Standard & non-standard cells, Redox applications to corrosion.
3. Inorganic chemistry: P Block and periodicity, Transition metals and their applications.
4. Introducing organic chemistry: Naming and drawing chemical structures.
5. Alkenes, alkynes and the electrophilic addition mechanism.
6. Aromatic compounds and electrophilic substitution mechanism.
7. Acids and bases in organic chemistry
8. Alkyl halides and nucleophilic substitution mechanism.
9. Functional Group Transformations
10. Polymers: synthetic and natural

A Timetable of Topics will be available from the eLearning site in week 1 of session.
Section B: Assessment

Assessment Summary

<table>
<thead>
<tr>
<th>Assessment Item</th>
<th>Form of Assessment</th>
<th>Due Date</th>
<th>Return/Feedback Due Dates</th>
<th>Weighting</th>
</tr>
</thead>
<tbody>
<tr>
<td>Assessment 1</td>
<td>Laboratory Reports</td>
<td>End of each Lab class</td>
<td>next Lab class</td>
<td>20%</td>
</tr>
<tr>
<td>Assessment 2</td>
<td>Moodle Online Quiz</td>
<td>Friday 10:00pm of Weeks 4, 6, 9, 11, 13.</td>
<td>on submissions</td>
<td>10%</td>
</tr>
<tr>
<td>Assessment 3</td>
<td>Mid-term Test</td>
<td>TBA</td>
<td>Lab class Week 10</td>
<td>10%</td>
</tr>
<tr>
<td>Assessment 4</td>
<td>Tutorial Preparation and Participation</td>
<td>Week 2-13</td>
<td>N/A</td>
<td>10%</td>
</tr>
<tr>
<td>Assessment 5</td>
<td>Final Examination</td>
<td>UOW Exam Period</td>
<td>Release of Results</td>
<td>50%</td>
</tr>
<tr>
<td><strong>Total Marks</strong></td>
<td></td>
<td></td>
<td></td>
<td><strong>100%</strong></td>
</tr>
</tbody>
</table>

Details of Assessment Tasks

Assessment tasks will be marked using explicit criteria that will be provided to students prior to submission.

**Assessment 1**
Laboratory Practical Reports
Due date: End of each laboratory class (= practicals)
Weighting: 20%
Submission: Submit a hardcopy to your tutor/demonstrator in class
Type of Collaboration: Individual and group work, note individual assessment
Length: 3-4 pages done in class time
Details: Templated practical report (template given in subject handbook) and questions submitted at the end of that practical. Returned marked with feedback at the next practical.
Style and format: (optional)
Subject Learning Outcomes: See each practical in Subject Handbook
Marking Criteria: Refer to Subject Handbook – Laboratory Reports

**Assessment 2**
Moodle Online Quiz
Due date: Friday 10:00pm of weeks 4, 6, 9, 11, 13.
Weighting: 10%
Submission: Submit the electronic quiz in Moodle
Type of Collaboration: Individual Assessment
Length: 10-15 MCQ’s or calculations
Details: Calculator Qs, A-E response questions – with feedback.
Subject Learning Outcomes: 1-6
Marking Criteria: Refer to Subject Handbook – online quiz

**Assessment 3**
Mid-term test
Due date: WTBA
Weighting: 10%
Submission: Test papers and answers must be submitted at the conclusion of the test
Type of Collaboration: Individual Assessment
Length: 50 Minutes
Details: Short answer questions
Subject Learning Outcomes: 1-5
Marking Criteria: Refer to Subject Handbook – mid-term test
**Assessment 4**  
**Tutorial participation and online activities (= tutorials)**  
**Due date**  
Weekly: Weeks 2-13  
**Weighting**  
10%  
**Submission**  
Complete pre-Tutorial in Moodle, participate actively in the tutorial  
**Type of Collaboration**  
Individual and group work  
**Length**  
50 minutes  
**Details**  
Online pre-tutorial and participation in face-to-face tutorial  
**Subject Learning Outcomes**  
1-6  
**Marking Criteria**  
Refer to Subject Handbook – Tutorials

**Assessment 5**  
**Final Examination**  
**Due date**  
UOW Exam Period  
**Weighting**  
50%  
**Submission**  
Exam papers and answers must be submitted at the conclusion of the exam  
**Type of Collaboration**  
Individual Assessment  
**Length**  
3hr  
**Details**  
70 MCQ’s  
**Subject Learning Outcomes**  
1-6  
**Marking Criteria**  
Refer to Subject Handbook – Examinations

**Minimum Requirements for a Pass in this Subject**
To receive a clear pass in this subject a total mark of 50% or more must be achieved. In addition, failure to meet any of the minimum performance requirements is grounds for awarding a Technical Fail (TF) in the subject, even where total marks accumulated are greater than 50%.

The minimum performance requirements for this subject are:
- achieve a minimum of 40% in the final exam  
- attend a minimum of 8 of the 9 laboratory classes. But note absence without Academic Consideration granted merits 0/10 included in your averaged lab mark  
- achieve a minimum of 50% average in laboratory practical reports

**Minimum Student Attendance and Participation**
It is expected that students will allocate 12 hours per week to this subject, including any required class attendance, completion of prescribed readings and assessment tasks.

Student attendance at practicals is compulsory and students must attend at least 100% of classes. Absences will require the submission of an application for Academic Consideration via SOLS and the presentation of suitable documentation, for example a Medical Certificate, to Student Central as soon as practical. For further details about applying for academic consideration visit the Student Central webpage: [http://www.uow.edu.au/student/central/academicconsideration/index.html](http://www.uow.edu.au/student/central/academicconsideration/index.html)

Attendance at lectures is highly recommended and frequently correlates with successful final grades. Attendance at weekly tutorials carries grade implications, see Assessment 4 above.

**Scaling**
Scaling will occur in this subject in the form of wave or broken stick scaling methods. For more information refer to Assessment Guidelines – Scaling: [http://www.uow.edu.au/about/policy/UOW039331.html](http://www.uow.edu.au/about/policy/UOW039331.html)
Late Submission

Late submission of an assessment task without an approved extension of the deadline is not acceptable. If you are unable to submit an assessment due to extenuating circumstances (e.g. medical grounds or compassionate grounds), you can make an application of academic consideration. Not all circumstances qualify for academic consideration. For further details about applying for academic consideration visit the Student Central webpage: http://www.uow.edu.au/student/central/academicconsideration/index.html

Late Submission Penalty – at 10%

Late submission of an assessment task without an approved extension of the deadline is not acceptable. Marks will be deducted for late submission at the rate of 10% of the total possible marks for that particular assessment task per day. This means that if a piece of work is marked out of 100, then the late penalty will be 10 marks per day (10% of 100 possible marks per day). The formula for calculating the late penalty is the total possible marks x 0.10 x number of days late. For the purposes of this policy a weekend (Saturday and Sunday) will be regarded as two days.

For example:
- Student A submits an assessment which is marked out of 100. The assessment is submitted 4 days late. This means that a late penalty of 40 marks will apply (100 x 0.10 x 4). The assessment is marked as per normal out of 100 and is given a mark of 85/100, and then the late penalty is applied. The result is that the student receives a final mark of 45/100 for the assessment (85 (original mark) – 40 marks (late penalty) = 45/100 (final mark)).
- Student B submits a report which is marked out of 20. The report is submitted three days late. This means that a late penalty of 6 marks will apply ((20 x 0.10 x 3). The report is marked as per normal out of 20 and is given a mark of 15/20, and then the late penalty is applied. The result is that the student receives a final mark of 9/20 for the report (15 (original mark) – 6 marks (late penalty) = 9/20 (final mark)).

No marks will be awarded for work submitted after the assessment has been returned to the students (except where a particular assessment task is undertaken by students at different times throughout the session, but where the assessment is based on experiments or case studies specific to a student). Notwithstanding this, students must complete all assessment tasks to a satisfactory standard and submit them, regardless of lateness or loss of marks, where submission is a condition of satisfactorily completing the subject.

Supplementary Assessments

Supplementary assessment may be offered to students whose performance in this subject is close to that required to pass the subject, and are otherwise identified as meriting an offer of a supplementary assessment. The precise form of supplementary assessment will be determined at the time the offer of a supplementary assessment is made.

Students can log on to SOLS and click on the link titled “Supplementary Assessment” to view any applicable offers or use the following link; http://www.uow.edu.au/student/exams/suppassess/index.html

System of Referencing Used for Written Work

The Author-Date (Harvard) referencing system should, unless otherwise specified for a particular assessment (check Details of Assessment Tasks), be utilised. A summary of the Harvard system can be accessed on the Library website at: http://uow.libguides.com/refcite

Submission of Assessments

Refer to the submission requirements under the details of the individual assessments. Students should ensure that they receive a receipt acknowledging submission. Students will be required to produce this in the event that an assessment task is considered to be lost. Students are also expected to keep a copy of all their submitted assessments in the event that re-submission is required.
Assessment Return

Students will be notified when they can collect or view their marked assessment. In accordance with University Policy marked assessments will usually only be held for 21 days after the declaration of marks for that assessment.
Section C: General Advice

Students should refer to the Faculty of Science, Medicine and Health website for information on policies, learning and support services and other general advice.

Student Consultation and Communication

University staff receive many emails each day. In order to enable them to respond to your emails appropriately and in a timely fashion, students are asked to observe basic requirements of professional communication.

Please ensure that you include your full name and student number and identify your practical class or tutorial group in your email so that staff know who they are communicating with and can follow-up personally where appropriate.

Consider what the communication is about

- Is your question addressed elsewhere (e.g. in the subject outline or, on the eLearning site)?
- Is it something that is better discussed in person or by telephone? This may be the case if your query requires a lengthy response or a dialogue in order to address. If so, see consultation times above and/or schedule an appointment.
- Are you addressing your request to the most appropriate person?

Specific email subject title to enable easy identification of issue

- Identify the subject code of the subject you are enquiring about (as staff may be involved in more than one subject) put this in the email subject heading. Add a brief, specific query reference after the subject code where appropriate.

Professional courtesy

- Address the staff member appropriately by name (and formal title if you do not yet know them).
- Use full words (avoid ‘text-speak’ abbreviations), correct grammar and correct spelling.
- Be respectful and courteous.
- Allow 3 – 4 working days for a response before following up. If the matter is legitimately urgent, you may wish to try telephoning the staff member (and leaving a voicemail message if necessary) or inquiring at the School Office.

eLearning Space

This subject has materials and activities available via eLearning. To access eLearning you must have a UOW user account name and password, and be enrolled in the subject. eLearning is accessed via SOLS (student online services). Log on to SOLS and then click on the eLearning link in the menu column. For information regarding the eLearning spaces please use the following link: https://www.uow.edu.au/student/learningcoop/index.html

Use of Internet Sources

Students are able to use the Internet to access the most current information on relevant topics and information. Internet sources should only be used after careful critical analysis of the currency of the information, the role and standing of the sponsoring institution, reputation and credentials of the author, the clarity of the information and the extent to which the information can be supported or ratified by other authoritative sources.
Lecture, Tutorial, Laboratory Times

On campus
All timetable information is subject to variation. Check latest timetabling information on the ‘Current Student’ webpage on UOW website or log into SOLS to view your personal timetable prior to attending classes.
Timetable information can be accessed from

Key University Dates can be accessed from

Extraordinary Changes for the Subject after Release of the Subject Outline
In extraordinary circumstances the provisions stipulated in this Subject Outline may require amendment after the Subject Outline has been distributed. All students enrolled in the subject must be notified and have the opportunity to provide feedback in relation to the proposed amendment, prior to the amendment being finalised.

Learning Analytics
Data on student performance and engagement (such as Moodle and University Library usage, task marks, use of SOLS) will be available to the Subject Coordinator to assist in analysing student engagement, and to identify and recommend support to students who may be at risk of failure. If you have questions about the kinds of data the University uses, how we collect it, and how we protect your privacy in the use of this data, please refer to
https://uow.edu.au/dvce/ltc/analytics/

The Assessment Quality Cycle
The Assessment Quality Cycle provides a level of assurance that assessment practice across the University is appropriate, consistent and fair.

Assessment Quality Cycle Activities are undertaken to contribute to the continuous improvement of assessment and promote good practices in relation to the:
- design of the assessment suite and individual assessment tasks;
- marking of individual assessment tasks;
- finalisation of subject marks and grades; and
- review of the subject prior to subsequent delivery

Copies of student work may be retained by the University in order to facilitate quality assurance of assessment processes.

Academic Integrity Policy
The full policy on Academic Integrity Policy is found in the Policy Directory on the UOW website.

“The University’s Academic Integrity Policy, Faculty Handbooks and subject guides clearly set out the University’s expectation that students submit only their own original work for assessment and avoid plagiarising the work of others or cheating. Re-using any of your own work (either in part or in full) which you have submitted previously for assessment is not permitted without appropriate acknowledgement or without the explicit permission of the Subject Coordinator. Plagiarism can be detected and has led to students being expelled from the University.
The use by students of any website that provides access to essays or other assessment items (sometimes marketed as ‘resources’), is extremely unwise. Students who provide an assessment item (or provide access to an assessment item) to others, either directly or indirectly (for example by uploading an assessment item to a website) are considered by the University to be intentionally or recklessly helping other students to cheat. Uploading an assessment task, subject outline or other course materials without express permission of the university is considered academic misconduct and students place themselves at risk of being expelled from the University.”

**Student Academic Complaints Policy (Coursework or Higher Degree Research)**

In accordance with the Coursework Student Academic Complaints Policy, a student may request an explanation of a mark for an assessment task or a final grade for a subject consistent with the student’s right to appropriate and useful feedback on their performance in an assessment task. Refer to the Coursework Student Academic Complaints Policy for further information.

**Student Support Services and Facilities**

Students can access information on student support services and facilities at the following link. This includes information on “Academic Support”, “Starting at University, “Help at University” as well as information and support on “Careers and Jobs”. [http://www.uow.edu.au/student/services/index.html](http://www.uow.edu.au/student/services/index.html)

**Student Etiquette**

Guidelines on the use of email to contact teaching staff, mobile phone use in class and information on the university guide to eLearning ‘Netiquette’ can be found at [https://www.uow.edu.au/student/learningcoop/software/emailetiquette/index.html](https://www.uow.edu.au/student/learningcoop/software/emailetiquette/index.html)
# UOW Grade Descriptors

The University of Wollongong Grade Descriptors are general statements that describe student performance at each of the University's grade levels.

<table>
<thead>
<tr>
<th>Grade</th>
<th>Mark %</th>
<th>Descriptor</th>
</tr>
</thead>
</table>
| High Distinction HD | 85-100 | A high distinction grade (HD) is awarded for performance that provides evidence of an outstanding level of attainment of the relevant subject learning outcomes, demonstrating the attributes of a distinction grade plus (as applicable):  
• consistent evidence of deep and critical understanding  
• substantial originality and insight in identifying, generating and communicating competing arguments, perspectives or problem-solving approaches  
• critical evaluation of problems, their solutions and their implications  
• use of quantitative analysis of data as the basis for deep and thoughtful judgments, drawing insightful, carefully qualified conclusions from this work  
• creativity in application as appropriate to the discipline  
• eloquent and sophisticated communication of information and ideas in terms of the conventions of the discipline  
• consistent application of appropriate skills, techniques and methods with outstanding levels of precision and accuracy  
• all or almost all answers correct, very few or none incorrect |
| Distinction D | 75-84  | A distinction grade (D) is awarded for performance that provides evidence of a superior level of attainment of the relevant subject learning outcomes, demonstrating the attributes of a credit grade plus (as applicable):  
• evidence of integration and evaluation of critical ideas, principles, concepts and/or theories  
• distinctive insight and ability in applying relevant skills, techniques, methods and/or concepts  
• demonstration of frequent originality in defining and analysing issues or problems and providing solutions  
• fluent and thorough communication of information and ideas in terms of the conventions of the discipline  
• frequent application of appropriate skills, techniques and methods with superior levels of precision and accuracy  
• most answers correct, few incorrect |
| Credit C      | 65-74  | A credit grade (C) is awarded for performance that provides evidence of a high level of attainment of the relevant subject learning outcomes, demonstrating the attributes of a pass grade plus (as applicable):  
• evidence of learning that goes beyond replication of content knowledge or skills  
• demonstration of solid understanding of fundamental concepts in the field of study  
• demonstration of the ability to apply these concepts in a variety of contexts  
• use of convincing arguments with appropriate coherent and logical reasoning  
• clear communication of information and ideas in terms of the conventions of the discipline  
• regular application of appropriate skills, techniques and methods with high levels of precision and accuracy  
• many answers correct, some incorrect |
| Pass P        | 50-64  | A pass grade (P) is awarded for performance that provides evidence of a satisfactory level of attainment of the relevant subject learning outcomes, demonstrating (as applicable):  
• knowledge, understanding and application of fundamental concepts of the field of study  
• use of routine arguments with acceptable reasoning  
• adequate communication of information and ideas in terms of the conventions of the discipline  
• ability to apply appropriate skills, techniques and methods with satisfactory levels of precision and accuracy  
• a combination of correct and incorrect answers |
| Fail F        | <50    | A fail grade (F) is given for performance that does not provide sufficient evidence of attainment of the relevant subject learning outcomes. |
| Technical Fail TF |        | A technical fail (TF) grade is given when minimum performance level requirements for at least one assessment item in the subject as a whole has not been met despite the student achieving at least a satisfactory level of attainment of the subject learning outcomes. |
| Satisfactory S |        | A satisfactory grade (S) is awarded for performance that demonstrates a satisfactory level of attainment of the relevant subject learning outcomes. |
| Unsatisfactory U |        | An unsatisfactory grade (U) is awarded for performance that demonstrates an unsatisfactory level of attainment of the relevant subject learning outcomes. |
| Excellent E   |        | An excellent grade (E) may be awarded, instead of a satisfactory grade (S), within subjects from the School of Medicine that have been completed with a consistent pattern of high standard of performance in all aspects of the subject. |

More details on UOW Grade descriptors can be found on the following link: [http://www.uow.edu.au/content/groups/public/@web/@gov/documents/doc/uow194941.pdf](http://www.uow.edu.au/content/groups/public/@web/@gov/documents/doc/uow194941.pdf)
University Policies

Students should be familiar with the following University policies:

a. Code of Practice – Teaching and Assessment

b. Code of Practice – Research, where relevant

c. Code of Practice – Honours, where relevant

d. Student Charter

e. Code of Practice – Student Professional Experience, where relevant

f. Academic Integrity and Plagiarism Policy

g. Student Academic Consideration Policy

h. Course Progress Policy

i. Academic Complaints Policy (Coursework and Honours Students)

j. Inclusive Language Policy

k. Workplace Health and Safety, where relevant

l. Intellectual Property Policy

m. IP Student Assessment of Intellectual Property Policy, where relevant

n. Policy on Ethical Objection by Students to the Use of Animal and Animal Products in Coursework Subjects, where relevant

o. Human Research Ethics Guidelines, where relevant

p. Animal Research Guidelines, where relevant

q. Student Conduct Rules and accompanying Procedures or Research Misconduct Policy for research students
<table>
<thead>
<tr>
<th>Version Control</th>
<th>Release Date</th>
<th>Author/Reviewer</th>
<th>Approved By</th>
<th>Amendment</th>
</tr>
</thead>
<tbody>
<tr>
<td>1</td>
<td>20190708</td>
<td>Glennys Obrien - Subject Coordinator</td>
<td>Sonia Losinno – Learning and Teaching Officer</td>
<td>Final Spring 2019 Subject Outline</td>
</tr>
<tr>
<td></td>
<td></td>
<td></td>
<td></td>
<td></td>
</tr>
</tbody>
</table>